Chemistry Courses (CHEM)

CHEM Courses

This is a list of courses with the subject code CHEM. For more information, see Chemistry (College of Liberal Arts and Sciences) in the Catalog.

CHEM:0500 Review of Chemistry Fundamentals 0 s.h.
Self-paced course to prepare for and improve success in CHEM:1110; students acquire an appropriate mathematics background and a sound understanding of some fundamentals of chemistry; use of ALEKS to test individual chemistry knowledge and then computerized adaptive learning software to fill gaps in knowledge.

CHEM:1000 First-Year Seminar 1-2 s.h.
Small discussion class taught by a faculty member; topics chosen by instructor; may include outside activities (e.g., films, lectures, performances, readings, visits to research facilities). Requirements: first- or second-semester standing.

CHEM:1050 Chemistry of Our World 3 s.h.
Nonmathematical exploration of selected areas of technology; basic science background, current technological applications, implications for society; for non-science majors. GE: Natural Sciences without Lab.

CHEM:1060 Technology and Society Laboratory 1 s.h.
Laboratory for CHEM:1050; demonstrations, student experiments. GE: Natural Sciences Lab only.

CHEM:1070 General Chemistry I 3 s.h.
Atomic structure, chemical bonds, mole relations, stoichiometry, states of matter, acids and bases, reaction rates, electrochemistry, nuclear chemistry. GE: Natural Sciences without Lab.

CHEM:1080 General Chemistry II 3 s.h.
Organic chemistry and biochemistry. GE: Natural Sciences without Lab.

CHEM:1090 Supplemental Chemistry Lab 1 s.h.
Lab techniques, elementary synthesis, measurement, analysis, case-study lectures and experiments; safety glasses, appropriate dress, compliance with laboratory safety protocols required.

CHEM:1100 Chemistry in Industry and the Economy 3 s.h.
Atomic structure, chemical bonding, acid and bases, polymers, pharmaceutics, DNA, proteins, and basic economics. GE: Natural Sciences without Lab.

CHEM:1110 Principles of Chemistry I 4 s.h.
Chemical bonding and chemical reactions; atomic and molecular structure, chemical equations, stoichiometry, gases, liquids, thermodynamics of phase changes, solutions, equilibrium, acids, bases, pH, elementary organic chemistry; the solid state, including modern materials; lecture, discussion, laboratory. Prerequisites: ALEKS score of 55 or higher or MPT Level 3 score of 9 or higher or MATH:1005 with a minimum grade of C- or MATH:1010 with a minimum grade of C- or MATH:1020 with a minimum grade of C- or MATH:1340 with a minimum grade of C- or MATH:1380 with a minimum grade of C- or MATH:1440 with a minimum grade of C- or MATH:1460 with a minimum grade of C- or MATH:1550 with a minimum grade of C- or MATH:1850 with a minimum grade of C-. Recommendations: Chemistry Diagnostic Test score of 16. GE: Natural Sciences with Lab.

CHEM:1120 Principles of Chemistry II 4 s.h.
Continuation of CHEM:1110; colligative properties of solutions, chemical thermodynamics, electrochemistry, chemical kinetics, chemical bonding, aspects of industrial chemistry, nuclear chemistry; lecture, discussion, laboratory. Prerequisites: CHEM:1110 with a minimum grade of C-. GE: Natural Sciences with Lab.

CHEM:1160 Principles of Chemistry Lab 2 s.h.
Laboratory techniques. GE: Natural Sciences Lab only.

CHEM:2021 Fundamentals of Chemical Measurements 3 s.h.
Introduction to experimental and data analysis techniques used in performing quantitative chemical measurements; topics include titrations, spectrophotometry, potentiometry, chromatography, and statistical techniques for use in data processing and interpretation; laboratory. Prerequisites: CHEM:1120 with a minimum grade of C-. Requirements: chemistry major.

CHEM:2210 Organic Chemistry I 3 s.h.
Carbon-containing compounds; structure, stereochemistry, physical properties, reactivity, reaction mechanisms, synthesis; emphasis on alkanes, alkenes, alkynes, ethers, alcohols, and alkyl halides. Prerequisites: CHEM:1120 with a minimum grade of C-.

CHEM:2220 Organic Chemistry II 3 s.h.
Continuation of CHEM:2210; use of spectroscopic techniques to determine chemical structures; chemistry of carbonyl compounds, amines, aromatics, amino acids, carbohydrates, nucleosides. Prerequisites: CHEM:2210 with a minimum grade of C- or CHEM:2230 with a minimum grade of C-.

CHEM:2230 Organic Chemistry I for Majors 3 s.h.
Carbon-containing compounds; structure, stereochemistry, physical properties, reactivity, reaction mechanisms, synthesis; emphasis on alkanes, alkenes, alkynes, ethers, alcohols, alkyl halides, aromatics. Prerequisites: CHEM:1120 with a minimum grade of C-. Recommendations: chemistry, biochemistry, or chemical engineering major.

CHEM:2240 Organic Chemistry II for Majors 3 s.h.
Continuation of CHEM:2230; use of spectroscopic techniques to determine chemical structures; chemistry of carbonyl compounds, amines, ethers, amino acids, carbohydrates, and nucleosides. Prerequisites: CHEM:2210 with a minimum grade of C- or CHEM:2230 with a minimum grade of C-. Recommendations: chemistry, biochemistry, or chemical engineering major.
CHEM:2410 Organic Chemistry Laboratory 3 s.h.
Preparation, purification, identification, analysis of chemical compounds, principally organic compounds. Prerequisites: CHEM:1120 with a minimum grade of C- and (CHEM:2210 with a minimum grade of C- or CHEM:2230 with a minimum grade of C-). Corequisites: CHEM:2220 or CHEM:2240.

CHEM:2420 Organic Chemistry Laboratory for Majors 3 s.h.
Preparation, purification, identification, analysis of chemical compounds, principally organic compounds. Prerequisites: CHEM:1120 with a minimum grade of C- and CHEM:2210 with a minimum grade of C- or CHEM:2230 with a minimum grade of C-). Corequisites: CHEM:2220 or CHEM:2240.

CHEM:3110 Analytical Chemistry I 3 s.h.
Modern theory and practice; emphasis on chemical equilibria (acid-base chemistry, solubility, complexation) and electroanalytical chemistry (potentiometry, voltammetry, coulometry). Prerequisites: CHEM:1120 with a minimum grade of C-.

CHEM:3120 Analytical Chemistry II 3 s.h.
Modern theory and practice; emphasis on atomic and molecular spectroscopy, mass spectrometry, chemical separations. Prerequisites: CHEM:1120 with a minimum grade of C-.

CHEM:3250 Inorganic Chemistry 3 s.h.
Modern principles; emphasis on descriptive chemistry of the main group and transition elements, ionic and covalent chemical bonding theories, symmetry, inorganic stereochemistry. Prerequisites: CHEM:2210 with a minimum grade of C- or CHEM:2230 with a minimum grade of C-). Corequisites: CHEM:2220 or CHEM:2240.

CHEM:3430 Analytical Measurements 3 s.h.
Modern theory and practice of laboratory methods; emphasis on experimental techniques and data analysis in spectroscopy, chromatography, electrochemistry. Prerequisites: CHEM:2021 with a minimum grade of C- and (CHEM:3110 with a minimum grade of C- or CHEM:3120 with a minimum grade of C-). Recommendations: (PHYS:1511 or PHYS:1611) and (PHYS:1512 or PHYS:1612).

CHEM:3440 Physical Measurements 3 s.h.
Laboratory experience using advanced instrumental and computational methods to generate and analyze data relevant to modern physical chemistry. Prerequisites: CHEM:2021 with a minimum grade of C- and (CHEM:4431 with a minimum grade of C- or CHEM:4432 with a minimum grade of C-).

CHEM:3530 Inorganic Chemistry Laboratory 3 s.h.
Preparation and characterization of a variety of inorganic, organometallic, and coordination compounds of the main group and transition elements; emphasis on synthetic techniques, methods for characterization of inorganic species. Prerequisites: CHEM:3250 with a minimum grade of C- and (CHEM:2410 with a minimum grade of C- or CHEM:2420 with a minimum grade of C- and CHEM:2021 with a minimum grade of C-.

CHEM:3560 Advanced Methods in Chemical Research: Special Topics 1-3 s.h.
Introduction to advanced research methods.

CHEM:3994 Undergraduate Research 1-4 s.h.

CHEM:4000 Scientists and Writers 1 s.h.
Science communication and collaborative skills that are highly sought after by employers in STEM fields including pharmaceutical firms, biotech start-ups, and many others; these same skills essential for reporting on, writing about, or translating science in any area; studio-style format. Same as JMC:4000, WRIT:4002.

CHEM:4171 Advanced Analytical Chemistry 3 s.h.
Emphasis on fundamental aspects of electrochemistry, atomic and molecular spectroscopy, chemical separations.

CHEM:4261 Selected Topics in Chemistry 1-3 s.h.
CheM:4270 Advanced Inorganic Chemistry 3 s.h.
Modern principles, including crystal field/ligand field/molecular orbital theory, inorganic reaction mechanisms, coordination chemistry, bioinorganic chemistry, main group and transition metal organometallic chemistry, solid-state inorganic chemistry. Prerequisites: CHEM:3250 with a minimum grade of C-.

CHEM:4372 Advanced Organic Chemistry 3 s.h.
Basic concepts from perspectives of structure, mechanism, synthesis, stereochemistry. Prerequisites: CHEM:2220 with a minimum grade of C- or CHEM:2240 with a minimum grade of C-.

CHEM:4430 Principles of Physical Chemistry 3 s.h.
Principles of physical chemistry, its application to chemical equilibrium, phase changes and chemical equilibria; ideal and real gases; kinetic theory; surface absorption and electrochemistry; thermodynamics. Prerequisites: CHEM:1120 with a minimum grade of C- and (MATH:1460 with a minimum grade of C- or MATH:1550 with a minimum grade of C- or MATH:1850 with a minimum grade of C-). Recommendations: (PHYS:1511 or PHYS:1611) and (PHYS:1512 or PHYS:1612).

CHEM:4431 Physical Chemistry I 3 s.h.
Chemical thermodynamics and its application to chemical equilibrium, phase changes and chemical equilibria; ideal and real gases; kinetic theory; surface absorption and electrochemistry; thermodynamics. Prerequisites: CHEM:1120 with a minimum grade of C- and (MATH:1560 with a minimum grade of C- or MATH:1860 with a minimum grade of C-).

CHEM:4432 Physical Chemistry II 3 s.h.
Quantum mechanics and its application to atomic and molecular structure; determination of structure and bonding by various spectroscopic methods; chemical kinetics. Prerequisites: CHEM:1120 with a minimum grade of C- and (MATH:1560 with a minimum grade of C- or MATH:1860 with a minimum grade of C-). Recommendations: PHYS:1512 or PHYS:1612.

CHEM:4450 Synthesis and Measurement 3 s.h.
Laboratory investigation integrating synthesis and measurement techniques from inorganic, analytical, and physical chemistry; emphasis on modern applications of chemistry in biology, medicine, environmental science, catalysis, and materials science. Prerequisites: CHEM:4432 with a minimum grade of C- or CHEM:4430 with a minimum grade of C- or CHEM:4431 with a minimum grade of C- and (CHEM:2420 with a minimum grade of C- or CHEM:2430 with a minimum grade of C- and CHEM:3120 with a minimum grade of C- or CHEM:3110 with a minimum grade of C- and CHEM:3250 with a minimum grade of C- and CHEM:2021 with a minimum grade of C-.
CHEM:4480 Introduction to Molecular Modeling 3 s.h.
Theory and application of ab initio quantum mechanics, semiempirical molecular orbital theory, and molecular mechanics force fields to chemical research problems; underlying theory of these methods (with emphasis on ab initio theory) and their practical application to chemical problems; computational chemistry projects using modeling software. Prerequisites: CHEM:4220 with a minimum grade of C- or CHEM:4210 with a minimum grade of C-.
Recommnedations: CHEM:4432.

CHEM:4760 Radiochemistry: Energy, Medicine, and the Environment 3 s.h.
Fundamental theoretical concepts of radiochemistry and their application in energy, medicine, and environmental sectors. Prerequisites: CHEM:2210 with a minimum grade of C- or CHEM:2230 with a minimum grade of C-

CHEM:4850 Upstream Biotechnology Processes 2 s.h.
Introduction to fermentation, fermenter preparation, cell growth and medium requirements, inoculation, sampling, process termination, separation of cells, fermentation case study, enzyme activity, and biocatalysis. Same as PHAR:4850.

CHEM:4873 Atmospheric and Environmental Chemistry 3 s.h.
Fundamental chemical processes of importance in the atmosphere, soil, and water, with emphasis on kinetics and photochemistry of homogeneous and heterogeneous reactions, atmospheric structure and dynamics, global geochemical cycling, chemistry-climate relationships, environmental remediation strategies; experimental methods in field and laboratory studies.

CHEM:4875 Introduction to Polymer Chemistry 2-3 s.h.
Synthesis, structures, characterization, properties, and applications of polymers. Prerequisites: CHEM:4431 or CHEM:4432.

CHEM:5091 Graduate Chemistry Orientation 2-3 s.h.
Pedagogy, safety, and research issues relevant to advanced chemistry careers.

CHEM:5107 Electrochemistry 2-3 s.h.
Fundamental aspects, including mass transport and electron transfer, electrochemical methodology (e.g., voltammetry and potentiometry), determination of homogeneous and heterogeneous reaction mechanisms. Recommendations: CHEM:3110, CHEM:3120, and CHEM:4171.

CHEM:5108 Spectroscopy 3 s.h.
Principles of atomic and molecular absorption and emission spectroscopy in ultraviolet, visible, and infrared regions of the spectrum, including fluorescence, phosphorescence, Raman spectroscopy; applications to analytical problems, with emphasis on modern instrumentan and methodology. Recommendations: CHEM:3110, CHEM:3120, and CHEM:4171.

CHEM:5109 Separations 3 s.h.
Analytical separations: basic theory, practical applications, instrumentation, modern techniques (extraction, gas and liquid chromatography, capillary electrophoresis), and detection (mass spectrometry). Recommendations: CHEM:3110, CHEM:3120, and CHEM:4171.

CHEM:5110 Chemical Sensors 2 s.h.
Theory, practical limitations, analytical utility based on immobilized reagents with electrochemical, thermal, optical transduction mechanisms. Recommendations: CHEM:3110 and CHEM:3120, or CHEM:4171.

CHEM:5114 Chemical Systems Modeling 3 s.h.
Basic processes and techniques; these methods applied to systems relevant to students’ own research. Recommendations: CHEM:3110 or CHEM:3120 or CHEM:4171.

CHEM:5115 Biophotonics 3 s.h.

CHEM:5118 Nanomaterials 3 s.h.
Basic principles associated with nanoscience and nanotechnology; fabrication and synthesis, size dependent properties, characterization, applications of materials at nanometer length scales, recent technological breakthroughs in the field. Requirements: graduate standing or advanced undergraduate standing in engineering and science. Recommendations: knowledge of basic chemistry.

CHEM:5120 Electrochemistry of Polymer Films 1 s.h.
Use of electrochemical methods to characterize polymer and thin films; transport through polymer films and composites, electrochemistry of polymer films. Requirements: physical chemistry course.

CHEM:5150 Chemometrics 3 s.h.

CHEM:5190 Seminar: Analytical Chemistry 0-1 s.h.
CHEM:5199 Special Topics in Analytical Chemistry arr.
Content varies.

CHEM:5202 Coordination Chemistry and Spectroscopy 1.3 s.h.

CHEM:5203 Organometallic Chemistry 3 s.h.

CHEM:5204 Physical Methods in Inorganic Chemistry 2 s.h.
Application of physical methods to problems; recent developments; emphasis on magnetic resonance spectroscopy. Recommendations: CHEM:4270.

CHEM:5205 Bioinorganic Chemistry 2-3 s.h.
The role of metal ions in biology from an inorganic chemical perspective; emphasis on structure and mechanism for transition metal-containing metallo-enzymes.

CHEM:5206 Solid-State and Materials Chemistry 3 s.h.
Introduction to the chemical concepts of solid-state chemistry; focus on synthesis and characterization of various inorganic materials; structure/property relationships, real-world examples. Recommendations: CHEM:4270.

CHEM:5212 Mass Spectrometry 3 s.h.
Examination of mass spectrometry in terms of basic theory, instrumentation, qualitative and quantitative analysis, and its application to the environmental and biological sciences. Recommendations: CHEM:3110 or CHEM:3120.

CHEM:5290 Seminar: Inorganic and Chemical Education Research 0-1 s.h.
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<tr>
<th>Course Code</th>
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<th>Credits</th>
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<tr>
<td>CHEM:4432</td>
<td>Special Topics in Inorganic Chemistry</td>
<td>1-3 s.h.</td>
<td>Fundamentals and applied aspects of surface chemical processes; theories of molecular adsorption/desorption and surface complexation; kinetics; surface analysis and instrumentation; applications of surface chemistry in heterogeneous catalysis, heterogeneous environmental/ atmospheric processes, and materials chemistry. Recommendations: CHEM:4431.</td>
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<tr>
<td>CHEM:5438</td>
<td>Surface Chemistry and Heterogeneous Processes</td>
<td>3 s.h.</td>
<td>Advanced topics related to modern physical chemistry. Recommended: CHEM:4432 and MATH:1860.</td>
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<tr>
<td>CHEM:5490</td>
<td>Seminar: Physical and Environmental Chemistry</td>
<td>0-1 s.h.</td>
<td>Advanced topics relevant to modern physical chemistry. Recommended: CHEM:4432 and MATH:1860.</td>
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<tr>
<td>CHEM:5499</td>
<td>Physical Chemistry Topics</td>
<td>1-3 s.h.</td>
<td>Special topics related to chemistry education; topics vary.</td>
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<tr>
<td>CHEM:5599</td>
<td>Special Topics in Chemistry Education</td>
<td>3 s.h.</td>
<td>Special topics related to chemistry education; topics vary.</td>
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<tr>
<td>CHEM:5990</td>
<td>Research Seminar</td>
<td>0-1 s.h.</td>
<td>Presentation and discussion of research by invited presenters.</td>
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<tr>
<td>CHEM:6990</td>
<td>Research Seminar</td>
<td>0-1 s.h.</td>
<td>Presentation and discussion of thesis research for advanced degrees.</td>
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<td>CHEM:7270</td>
<td>Ethics in Chemical Sciences</td>
<td>1 s.h.</td>
<td>Scholarly integrity for being a responsible chemist on graduate-level research; introduction to infrastructure of scientific scholarship with emphasis on interacting with peers, funding agencies, industrial entities; responsible conduct in research in the context of creation of knowledge, dissemination of scientific findings, intellectual property, and conflict of interest; workshops to study cases in chemical research to illustrate the principles of scholarly integrity.</td>
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<tr>
<td>CHEM:7604</td>
<td>Ethics in Chemical Sciences for Postdocs</td>
<td>0 s.h.</td>
<td>Introduction to infrastructure of scientific scholarship; emphasis on interacting with peers, funding agencies, industrial entities; scholarly integrity for being a responsible chemist on graduate-level research; responsible conduct in research in context of creation of knowledge, dissemination of scientific findings, intellectual property, conflict of interest; workshop cases in chemical research that illustrate principles of scholarly integrity.</td>
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**CHEM:5299 Special Topics in Inorganic Chemistry**
Recommendations: CHEM:4270.

**CHEM:5321 Spectroscopic Methods in Organic Chemistry**

**CHEM:5326 Organic Reactions**
Survey of organic reactions used in contemporary organic synthesis; emphasis on C-C bond forming reactions, functional group interconversions, oxidations and reductions; mechanistic details of reaction types; innovations in catalytic and asymmetric organic reactions. Recommendations: CHEM:4372.

**CHEM:5328 Mechanisms of Organic Reactions**
Application of basic mechanistic concepts.

**CHEM:5329 Advanced Organic Synthesis**

**CHEM:5390 Seminar: Organic Chemistry**
0-1 s.h.

**CHEM:5399 Organic Chemistry Special Topics**
1,3 s.h. Recommendations: CHEM:4372.

**CHEM:5431 Statistical Thermodynamics I**
3 s.h. Fundamentals of classical thermodynamics and equilibria; ensembles; noninteracting systems; theory of phase transitions; Monte-Carlo methods; classical fluids; nonequilibrium systems. Recommendations: CHEM:4431.

**CHEM:5433 Quantum and Computational Chemistry**
3 s.h. Fundamental principles of quantum chemistry; angular momentum; approximation methods; theory of atomic and molecular electronic structure; applications of computational quantum mechanics to chemical systems. Corequisites: CHEM:4432, if not taken as a prerequisite.

**CHEM:5434 Molecular Spectroscopy**
3 s.h. Quantum mechanical theory of molecular spectroscopy; time-dependent perturbation theory, selection rules, line shapes; selected applications in microwave, vibrational (infrared and Raman), electronic, optical, and magnetic resonance spectroscopy. Recommendations: CHEM:5433.

**CHEM:5435 Chemical Kinetics**
3 s.h. Potential energy surfaces, transition state theory, diffusion limited rates, linear free energy relationships, isotope effects, solvent effects, RRKM theory; connection between experiment and various theories in the gas and solution phases; emphasis on assignment of experimental error to derived quantities. Recommendations: CHEM:4432.

**CHEM:5436 Electronic Structure and Informatics in Chemistry**
3 s.h. Basic principles of molecular electronic structure theory; molecular structure and reactivity; molecular orbital theory; density functional theory; introduction to informatics and data science; how calculations can be used to enhance experimental research projects. Recommendations: CHEM:4432. Same as IGPI:5436.